

Oxidation and Exfoliation in sCO₂ Power Cycles for Transformational Fossil Power Systems



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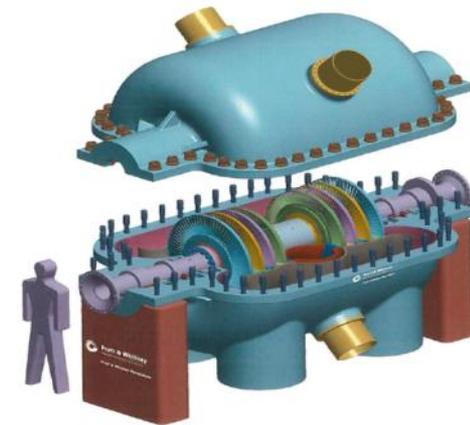
*5th International Symposium – Supercritical CO₂
Power Cycles
San Antonio, TX USA
March 29-31, 2016*

Project Objectives

■ Overall Objective

- Predict the oxidation/corrosion performance of structural alloys in high-temperature high-pressure supercritical CO₂ (sCO₂)
- Combine laboratory testing & computational modeling including unique attributes of sCO₂ heat exchangers to accomplish this goal

■ Materials for sCO₂ help Enable US DOE Program Goals for Future Transformational Power Systems



sCO₂ Power Turbine
(676 MW)

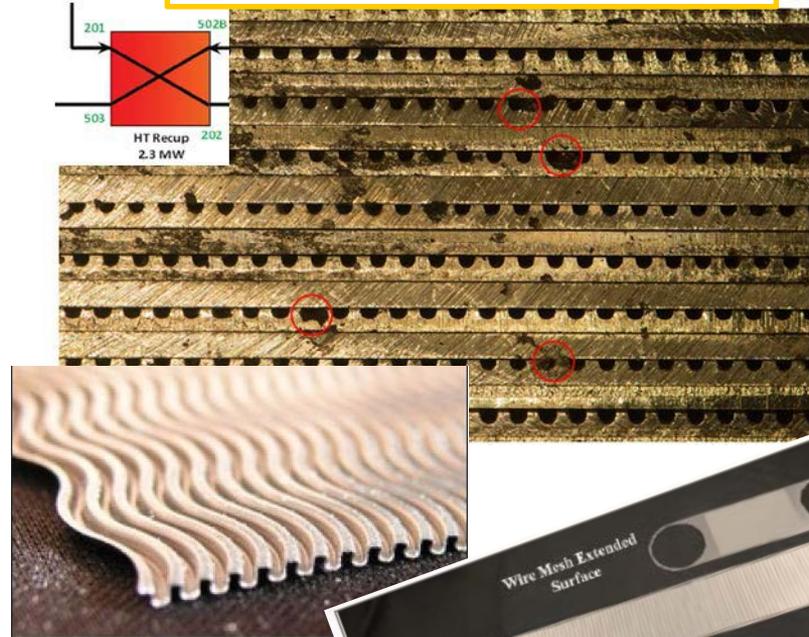
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Some configurations of the sCO₂ Brayton power cycle might achieve 100% carbon dioxide capture and zero emissions of conventional pollutants with little or no efficiency or capacity penalty.

Technology Challenges for sCO₂

- HX Expensive: '40% of plant cost'
- Small channels
- Large surface areas
- Materials considerations: thermal fatigue, creep (thin sections), brazing/diffusion bonding, corrosion/oxidation/carburization
- **Corrosion/Oxidation**
 - Closed cycle = build-up of impurities
 - **Open cycle = combustion products**
 - Long-term performance, pluggage, blockage, etc.

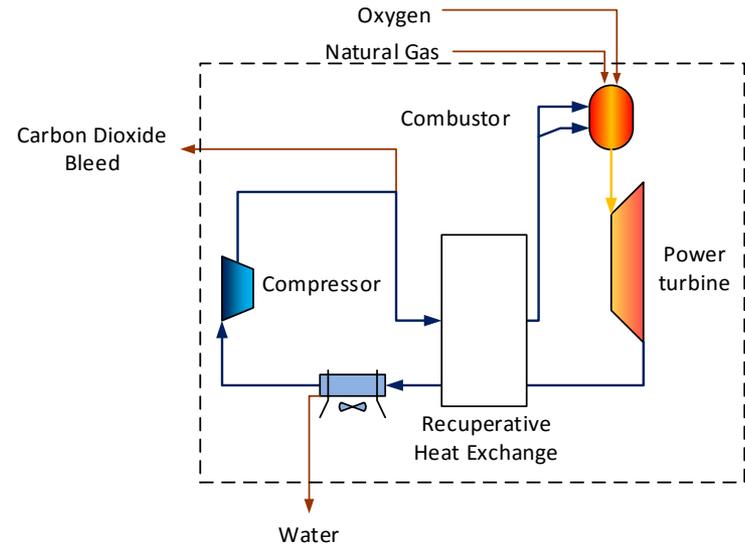
Compact Heat Exchanger Fouling (Sandia National Lab)



New Heat Exchanger Design (Brayton Energy)

Realistic sCO₂ conditions for semi-open Allam cycles

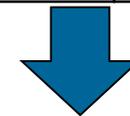
- **Survey of industry and current studies**
 - 700°C likely maximum temperature in heat-exchangers
- **Evaluation of impurities for nearest-term ‘open/direct-fired cycle’ – Allam Cycle**
 - H₂O, O₂, N₂, Ar, NO_x, SO_x, HCl
 - Mass-balance calculations for methane and cooled, raw syngas (checked against thermodynamic calculations)



Species	Composition (mol%)		
	Methane	Cooled raw coal syngas	Oxygen
CH ₄	100	1.0	
CO		39.0	
H ₂		28.3	
CO ₂		8.0	
H ₂ O		20.0	
N ₂ +Ar		2.0	0.5
H ₂ S		0.9	
HCl		0.02	
O ₂			99.5
LHV	912 BTU/scf	218 BTU/scf	



Component	Composition (mol%)			
	Methane		Cooled Raw Coal Syngas	
	Combustor Inlet	Turbine Inlet	Combustor inlet	Turbine Inlet
CO ₂	95	90	90	85
H ₂ O	250 ppm	5	250 ppm	5
N ₂ +Ar	1	1	9	9
O ₂	4	4	1	1
HCl				20 ppm
SO ₂				1,000 ppm



O₂ = 3.6 vol%, H₂O = 5.3 vol%

Scope of Laboratory sCO₂ Corrosion Tests

■ Conditions

- 600-750°C, 200 bar
- sCO₂
 - Commercially pure
 - Simulated semi-open cycle impurities (O₂ + H₂O)

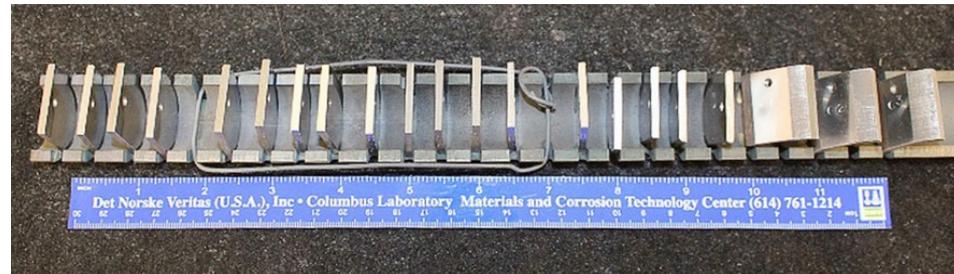
■ Materials

- Commercially available
- Code approved or Industrially relevant
- Focus on economics

■ Exposures

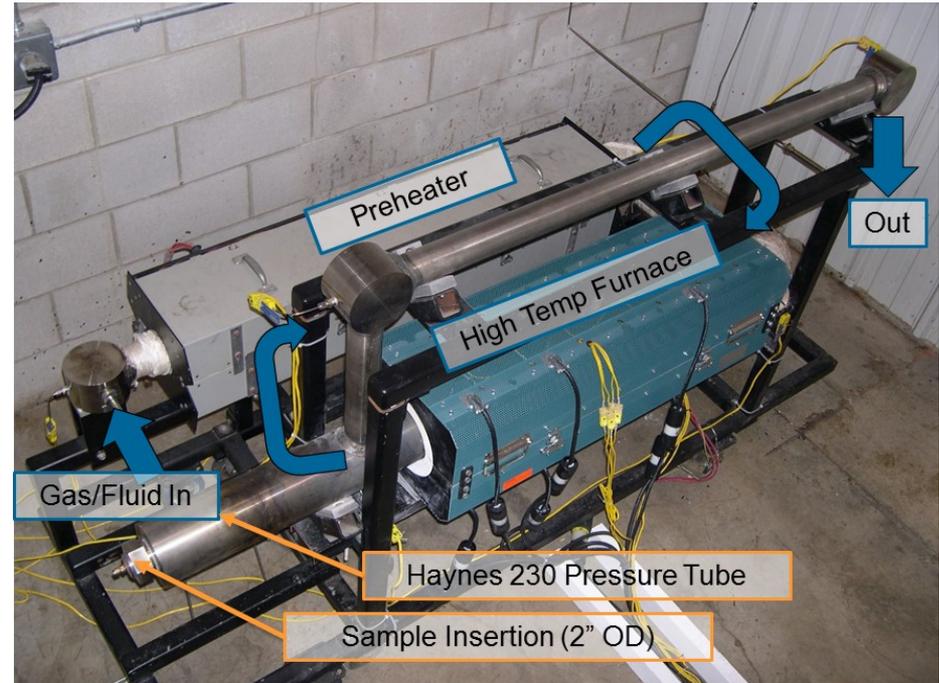
- 300 hours (Gr 91, 304H, 740H), 700°C
- 1,000 hours (all 7 alloys), 3 temperatures
- ≥3,000 hours (all 7 alloys), 1 temperature

Material Class	Alloys Selected		
Ferritic steels	Gr. 91 (8-9Cr)	VM12 (11-12Cr)	Crofer 22H (20Cr)
Austenitic stainless	304H (18Cr)	310HCbN (25Cr)	
Nickel-based	617 (20Cr, solid soln)	740H (25Cr, ppt. strengthnd)	

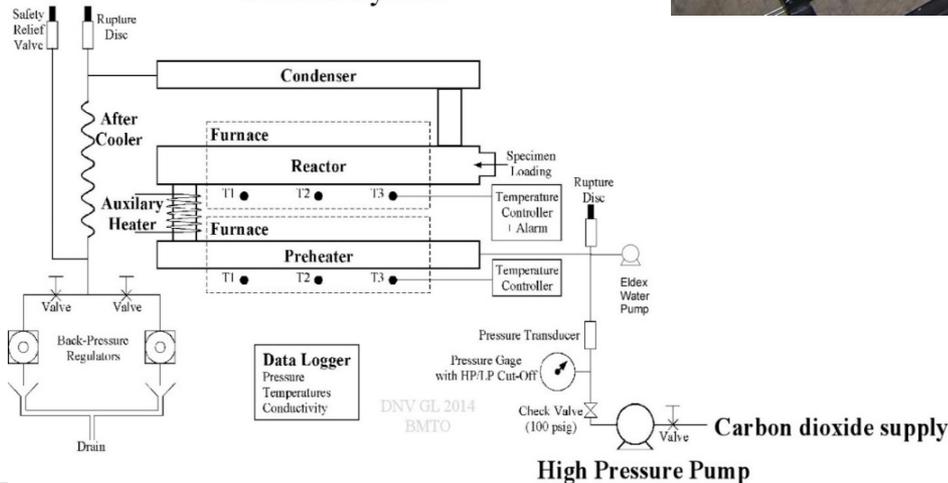


Laboratory Testing Facility (DNV-GL)

- High temperature and pressure (600-750°C, 200 bar)
- Existing test facility modified for sCO₂ to ensure safety
- Introduction of impurities (O₂, H₂O)
- **300-hour tests in sCO₂ with and without impurities completed successfully**



Supercritical Carbon Dioxide System

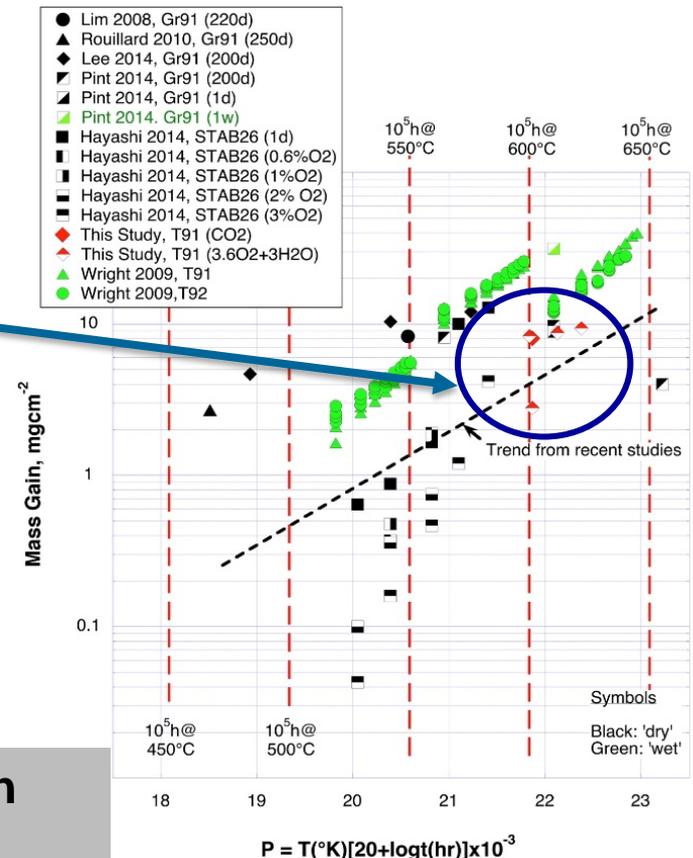


Comparison of Mass gain in sCO₂ and steam

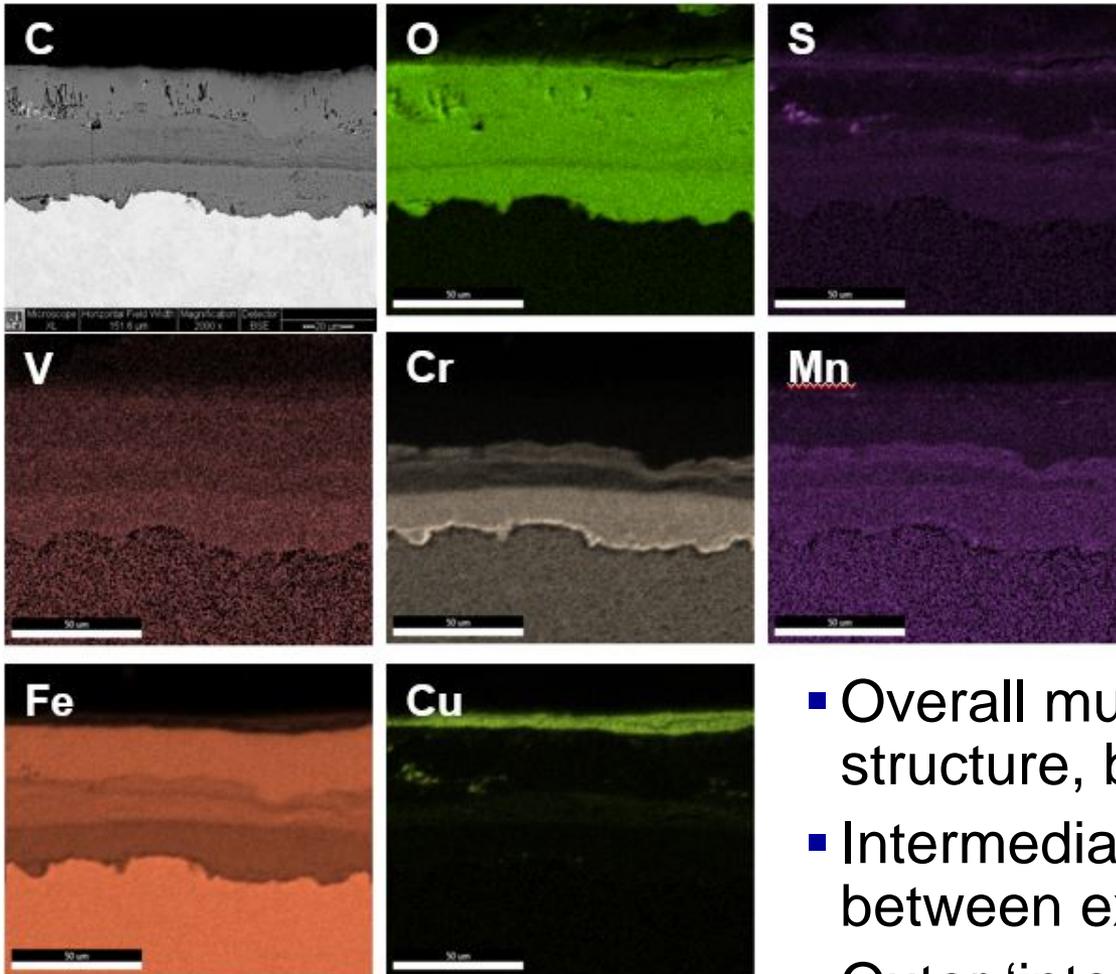
- Results from 300 hour test in pure sCO₂, 700°C, 200 bar Mass gains are similar to results in steam and other studies in sCO₂
- However, mass gain is not useful for evaluating oxide morphology and propensity for exfoliation**

Sample ID	Sample #	Weight gain	
		mg	mg/cm ²
T91	1	124.57	7.66
	2	143.47	8.82
	3	124.17	7.63
TP304H	1	4.53	0.28
	2	2.77	0.17
	3	3.97	0.24
740	1	3.13	0.19
	2	3.47	0.21
	3	3.80	0.23

300 hr mass gain data are consistent with assembled literature data



Grade 91 after 300 hour test in pure CO₂ at 200 bar, 700°C: Oxide Morphology



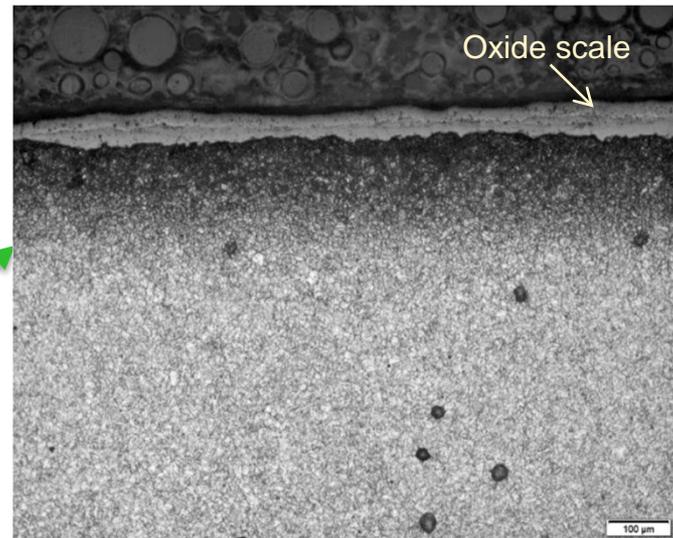
- Overall multi-layered scale structure, but
- Intermediate layers are present between expected L1 & L2
- Outer 'intermediate layer' contains Cr (level lower than in L1)

Evidence of Carburization on Gr 91 after 300 hour in pure sCO₂ at 200 bar, 700°C

- Decoration of etched Gr 91 microstructure
- Initial spot hardness measurements for carburization inconclusive
- More detailed characterization pursued

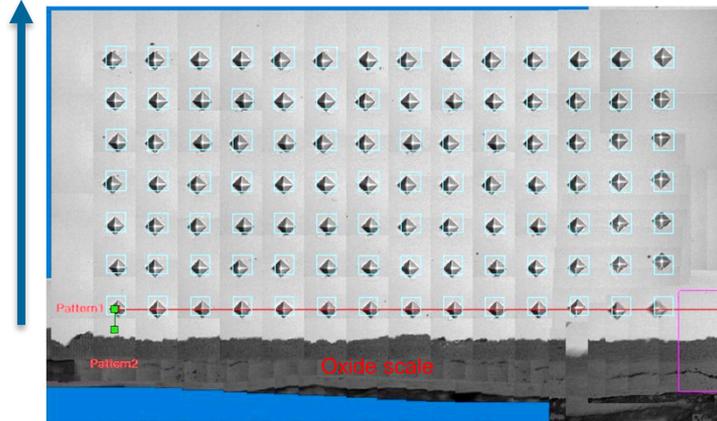
Hardness Spot Checks

Depth (microns)	Hardness (Vickers)
625	224
417	234
271	251
167	321
83	297

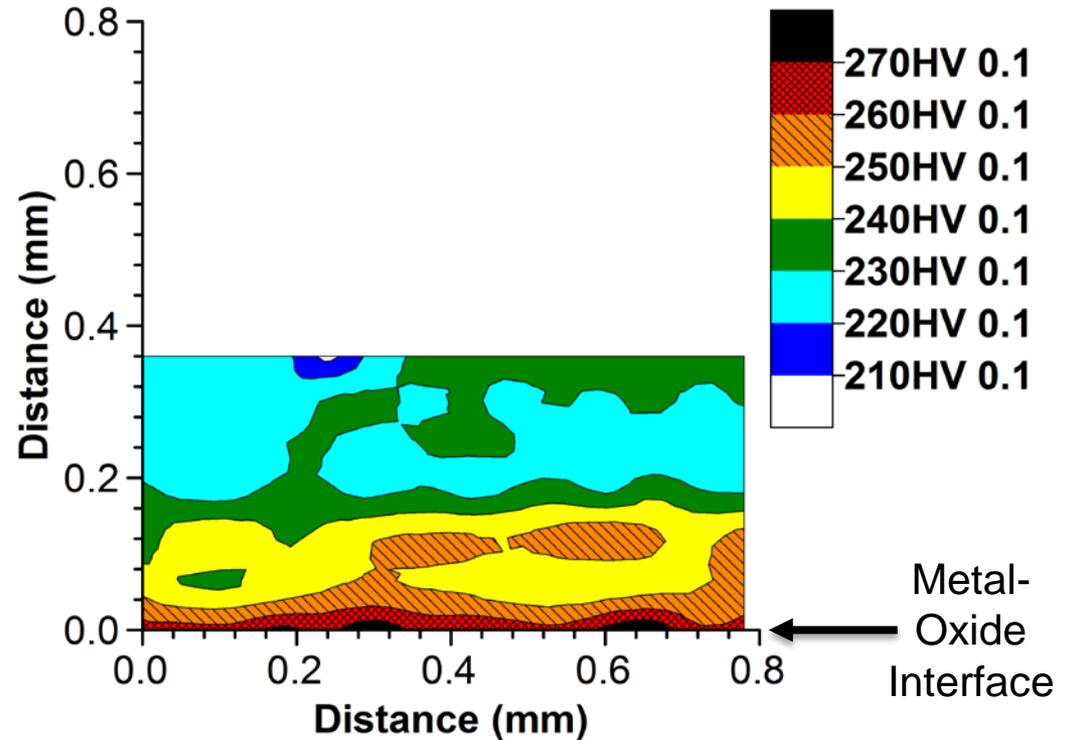


Carburization evident from micro-hardness measurements on Gr 91 in pure sCO₂ after 300 hour at 200 bar, 700°C

Depth into alloy

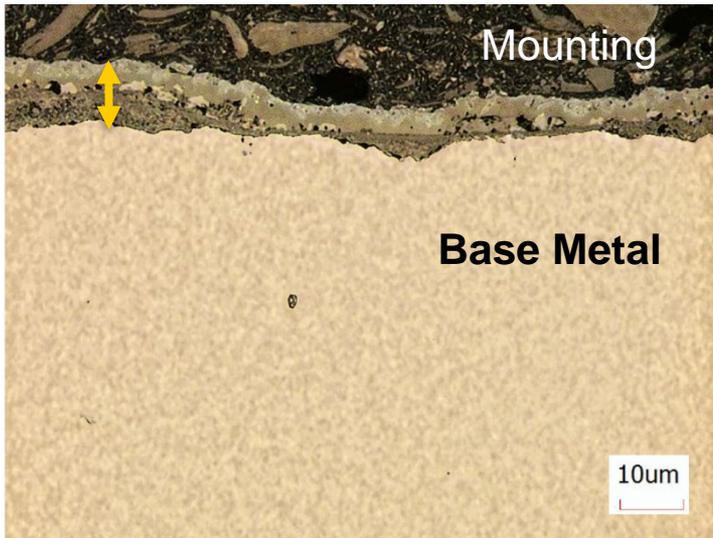


Hardness Map: 14x7=98indents



Automated hardness map shows harder near-surface region

Laser micrographs of 304H and 740H after 300 hours in pure CO₂ at 200 bar, 700°C



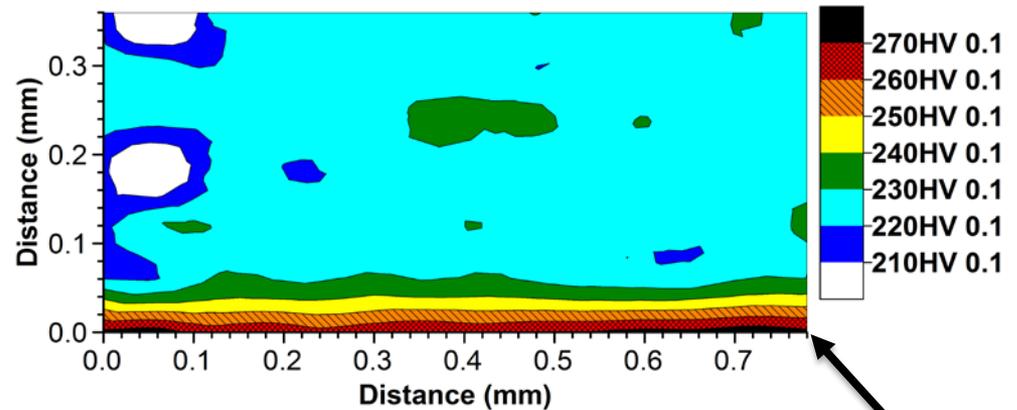
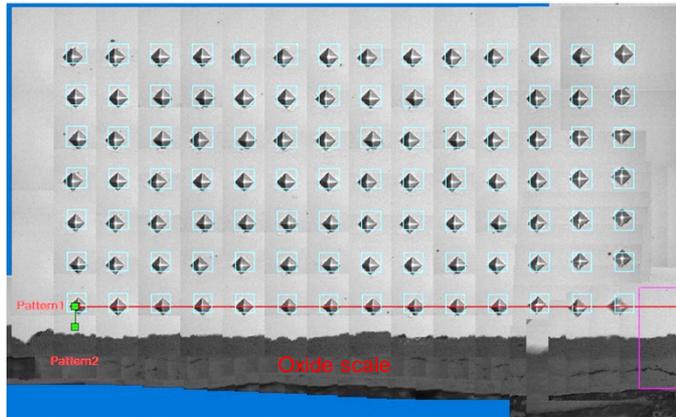
304H



740H

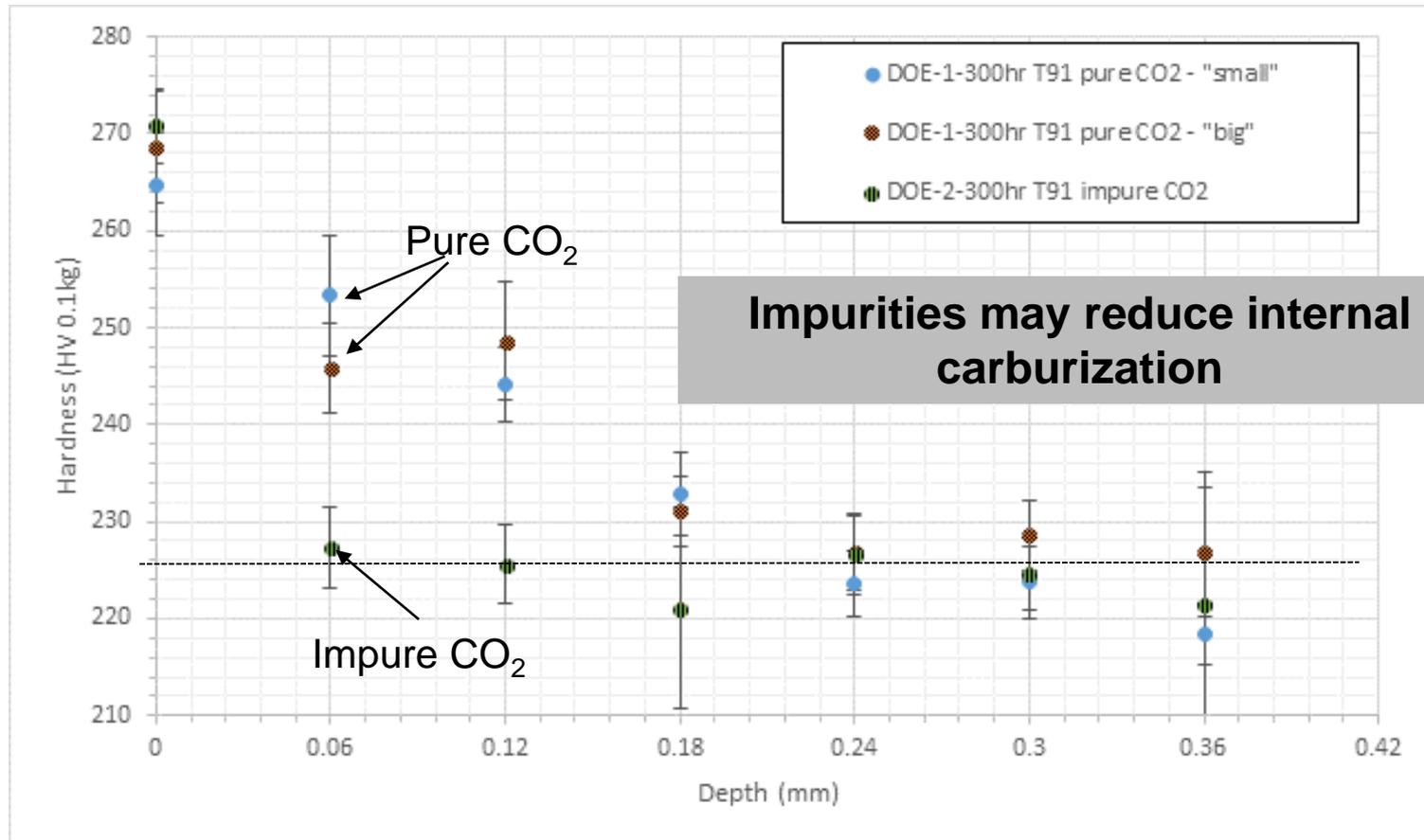
Very thin oxide layers on higher-Cr austenitic and nickel-based alloy at short-times

Hardness on Gr 91 after 300 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C



Behavior appears to have some dependence on impurity content of sCO₂

Hardness Profiles on Gr 91 after 300 hour test in pure and impure sCO₂ at 200 bar, 700°C

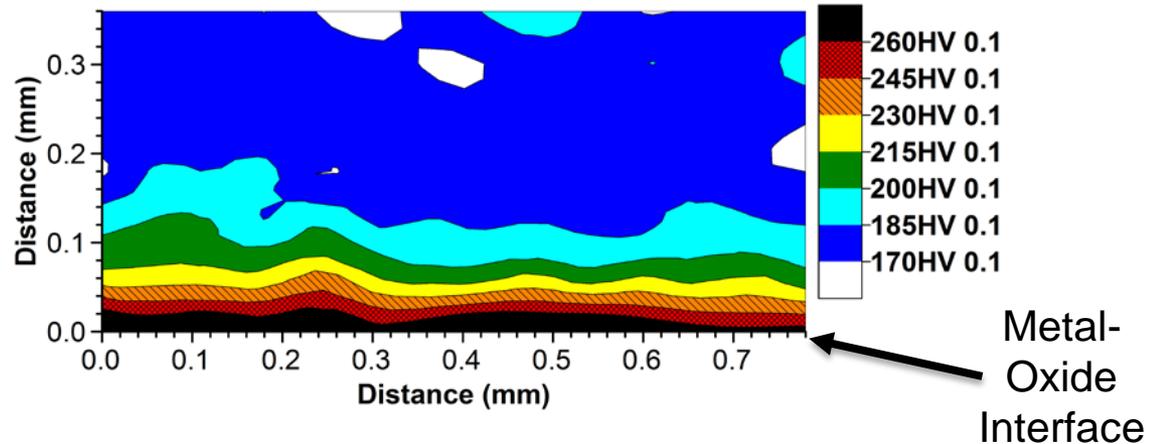


- Carburization depth >200 μm after 300 hours
- Can it lead to breakaway corrosion?

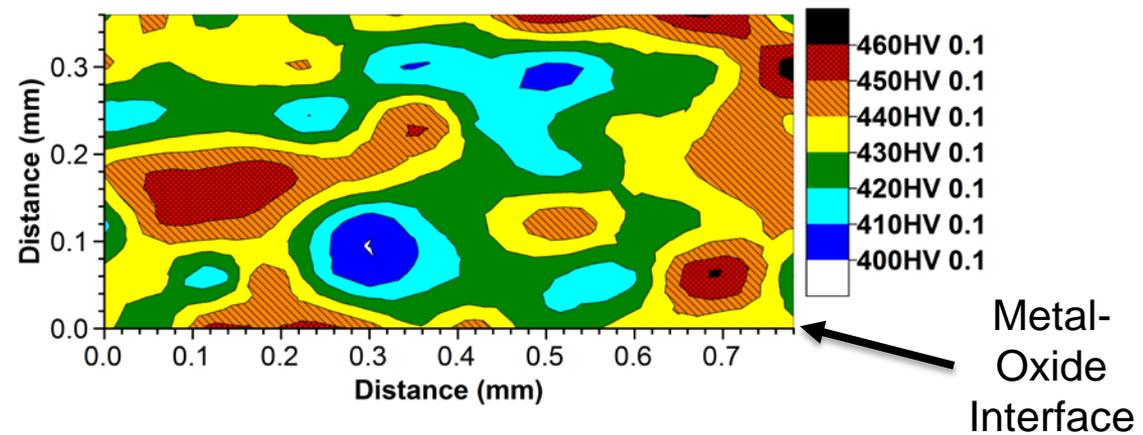
Hardness on 304H and Inconel 740H after 300 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C

Stainless Steel 304H

304H shows some hardness increase near the alloy surface – need to evaluate further (sample prep, carburization, other)

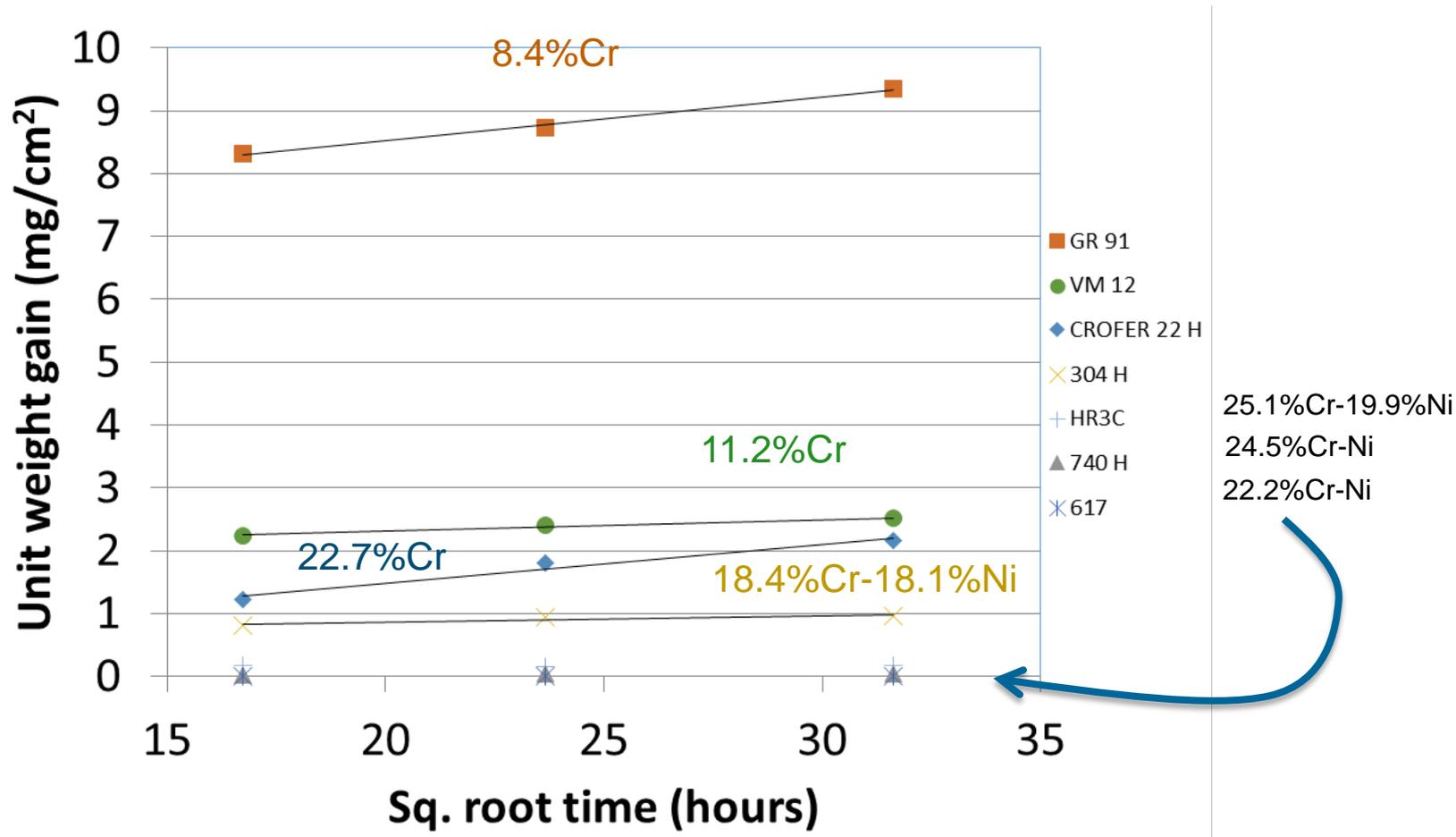


Nickel-Based Alloy 740H



740H shows no evidence of change in surface hardness with sCO₂ exposure

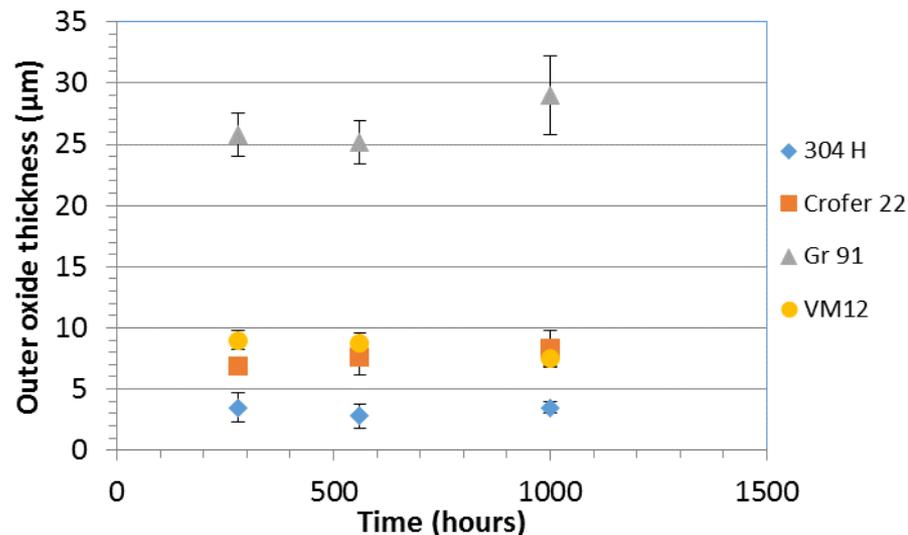
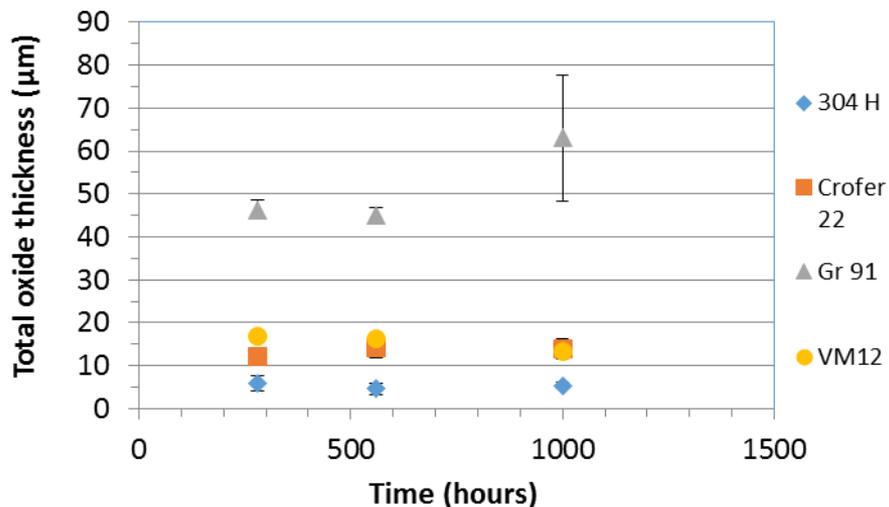
Mass Gain from 1000 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C



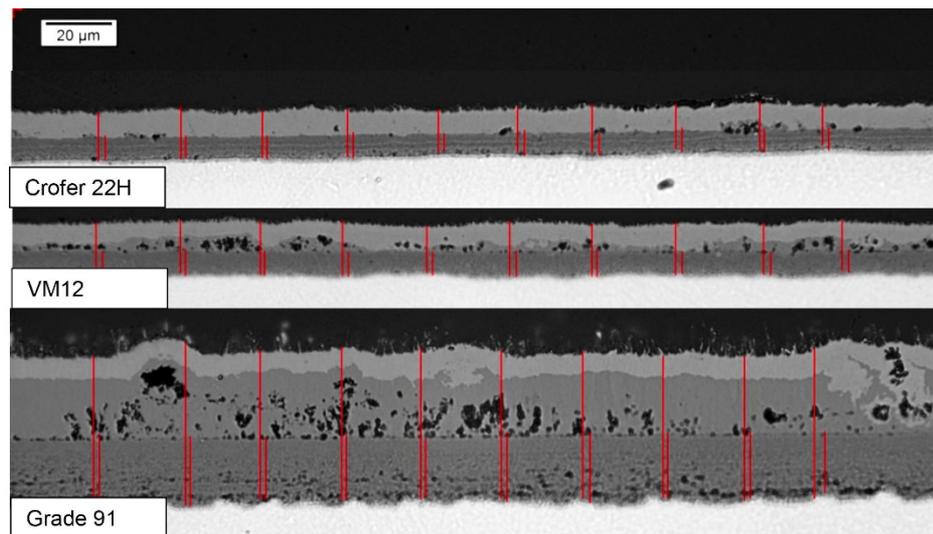
- Mass gain trend with alloy chromium content for Fe-based alloys
- 304H stainless steel has highest mass gain of austenitics
- Mass gains for HR3C, 617, and 740H are comparable

Comparison of Ferritic Alloys and 304H Stainless Steel

1,000 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C



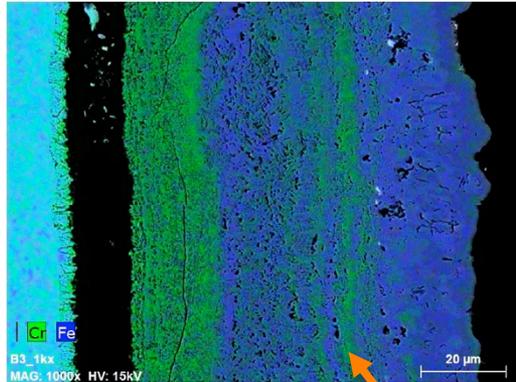
- All ferritic alloys form duplex scale structure at 700°C, even with ~23%Cr
- No exfoliation observed (yet)
 - EPRI models for steam predict exfoliation from Gr.91 at 200 to 400 microns total oxide thickness



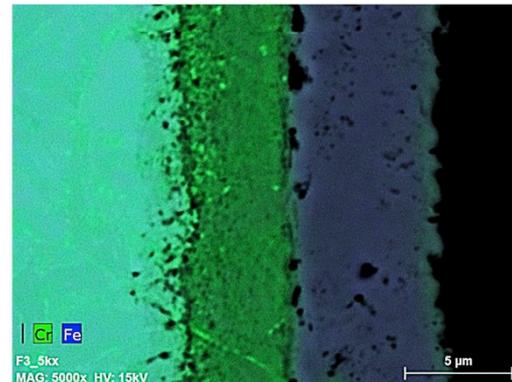
Comparison of Ferritic Alloys and 304H Stainless Steel

1,000 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C

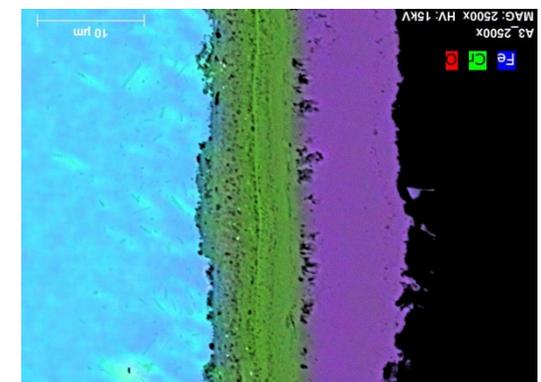
EDS Fe-Cr or Fe-Cr-O Maps Overlayed on SEM Images



Gr 91: 8.4%Cr

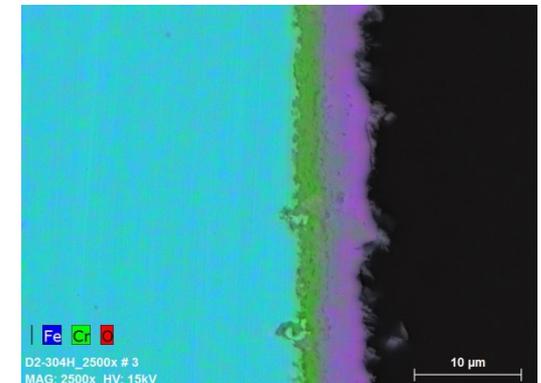


VM12: 11.2%Cr



Crofer 22: 22.7%Cr

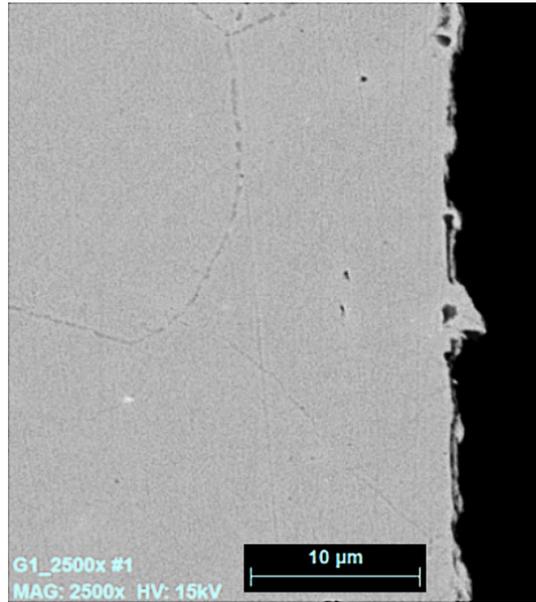
- Outer oxides are Fe-based
 - Gr. 91 continues to show **intermediate layer(s) showing Cr & Fe striations**
 - With exception of Gr. 91, oxide morphologies appear similar to those in steam
- No exfoliation observed, but
 - outer Fe-oxide (L2) growing on all alloys suggests eventual exfoliation
 - Voids already forming on L1/L2 interfaces on ferritic alloys--these are typical locations for scale failure



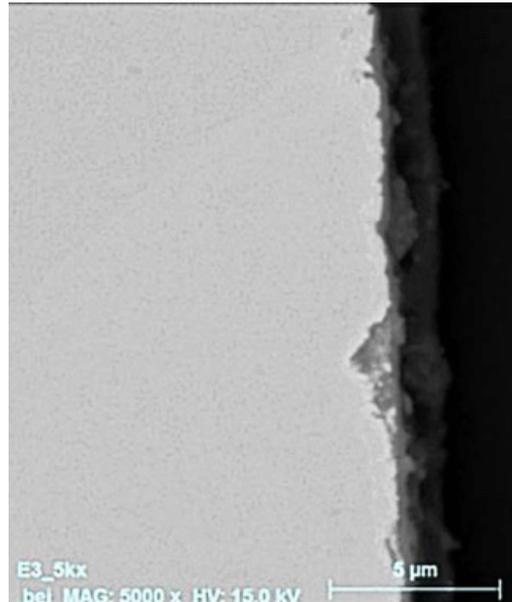
304 Stainless Steel:
18.4%Cr-18.1%Ni

High-Cr Stainless Steel and Nickel-Based Alloys

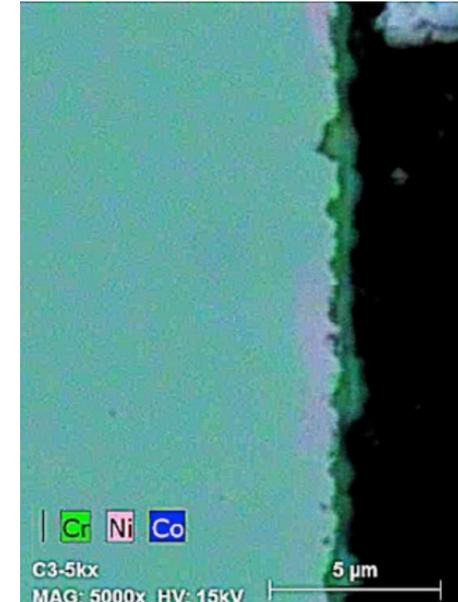
1,000 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C



HR3C: 25.1%Cr-19.9%Ni



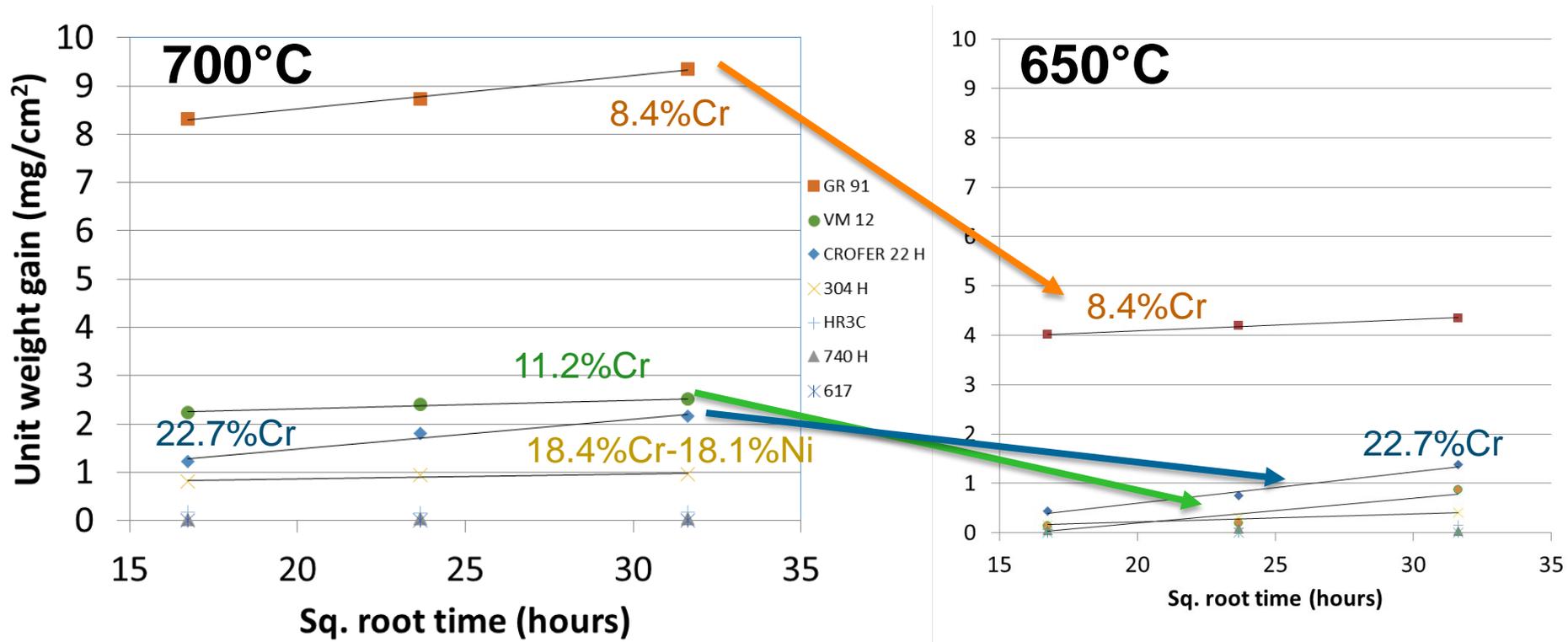
617: 22.2%Cr-Ni



740H: 24.5%Cr-Ni

- Very thin oxide scales
- Surface roughness > oxide thickness

Recent Data: Mass Gain from 1000 hour test in impure sCO₂ (3.6% O₂, 5.3% H₂O) at 200 bar, 700°C versus 650°C

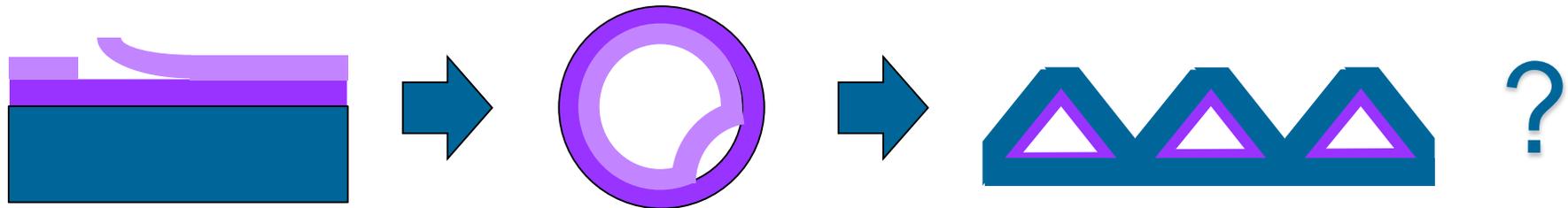
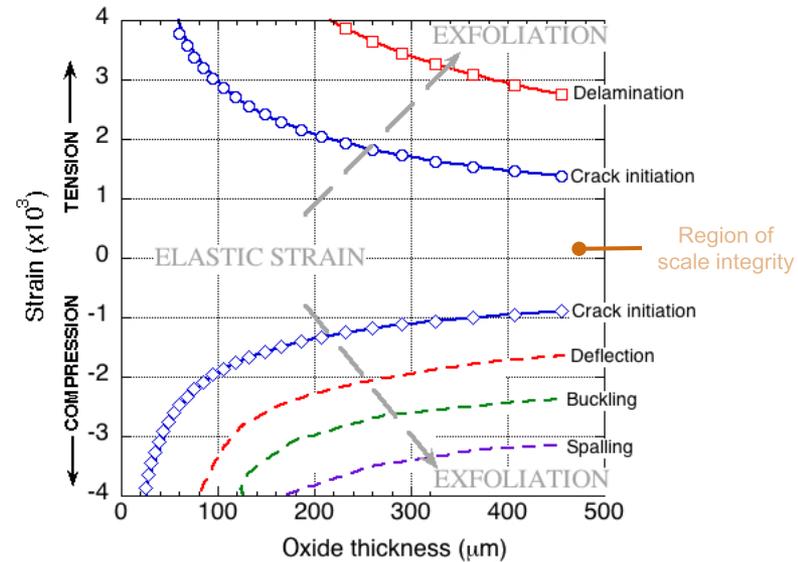


- Similar trends at 650°C to 700°C

- Weight gain reduced by ½ with 50°C reduction in temperature
- VM12 and Crofer 22H ranking inconsistent between tests
- Oxide scales being characterized

Unique consideration for oxide growth and exfoliation in small channel heat-exchangers

- Lab studies: isothermal oxide growth on small flat coupons
- Real world: heat-flux, stress from complex geometries
- Modeling:
 - EPRI-developed strain trajectory approach for steam tubes
 - Properties of sCO₂ and alloys collected
 - **Discussion with vendors on convex vs. concave surfaces – need to develop a generic modeling approach**



See presentation by Adrian Sabau for details

Test Matrix Progress

Description (Purpose)	Conditions	Test Status	Characterization
Rig Commissioning	Temperature monitoring & pressure	Complete	n/a
Short-term (compare to literature, impurity introduction)	700°C-300hr-Pure	Complete	100%
	700°C-300hr-Impure	Complete	100%
Test Program (develop oxide thickness kinetics and propensity for exfoliation)	700°C-1,000hr-Impure	Complete	Ongoing: ~300, 700, 1,000hr: mass gain, oxide thickness, morphology
	650°C-1,000hr-Impure	Complete	
	750°C-1,000hr-Impure	Ongoing	
Long-Term (Validate Models and test unique geometries)	700°C-3,000+hr-Impure	Discussions with vendors to test actual sub components	

Summary

- First project to address oxidation in open sCO₂ Allam cycle
 - Impurity concentrations have been determined via mass balance and thermodynamic calculations
- A new test rig has been assembled and 300-hour laboratory tests with and without impurities completed
 - 1,000-hour tests are progressing at 650-750°C
- Although short-term mass gains for alloys in sCO₂ and steam are similar, some differences in scale morphology
 - Intermediate layer between L1 & L2 with Cr striations in Gr. 91
 - Carburization identified on Gr 91 using hardness mapping; appears more severe in pure sCO₂
 - Possible carburization on stainless steels requires more investigation and longer-term test
- Effect of sCO₂ and geometry on oxidation to be evaluated through modeling [Separate Presentation]

Acknowledgements / Team

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- DOE/NETL
 - Vito Cedro, Project Manager
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 - John Shingledecker (Principal Investigator)
 - David Thimsen
 - Steve Kung
- DNV-GL
 - Brett Tossey
- WrightHT
 - Ian Wright
- Oak Ridge National Laboratory
 - Adrian Sabau

8th International Conference on Advances in Materials Technology for Fossil Power Plants

- **October 11-14, 2016**: Sheraton Algarve: Albufeira, Portugal
- **Materials for: *power steam boilers, steam turbine, gas turbines, and heat recovery steam generators***:
 - Plant economics, advanced designs, and field experience
 - New materials, advanced alloys, & material design concepts
 - Tubes, pipes, rotors/discs, casings
 - Creep, fatigue, toughness, creep-fatigue interaction
 - Microstructural evolution
 - Coatings, corrosion, claddings
 - Welding & fabrication



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Alloy Compositions for DOE sCO ₂ Corrosion Study (wt.%)							
	T91	VM12	Crofer 22H	304H	HR3C	617	740H
Al	0.01	0.01	0.01	<0.01	0.01	1.13	1.33
B	0.002	0.004	<0.001	0.001	0.001	0.002	0.001
Ce	-	-	-	<0.01	<0.01	<0.01	<0.01
Ca	<0.01	<0.01	<0.01	-	-	-	-
Co	0.02	1.47	0.02	0.22	0.08	11.44	20.28
Cr	8.39	11.2	22.71	18.42	25.13	22.19	24.53
Cu	0.09	0.08	0.01	0.18	0.03	0.03	0.01
Fe	-	-	-	70.33	52.39	1.55	0.12
La	<0.01	<0.01	0.06	-	-	-	-
Mn	0.44	0.39	0.43	1.8	1.19	0.09	0.26
Mo	0.93	0.36	0.01	0.22	0.1	9.5	0.32
Nb	0.06	0.03	0.5	0.01	0.44	0.06	1.49
Ni	0.13	0.36	0.26	8.13	19.85	53.31	50.04
P	0.014	0.015	0.018	0.028	0.015	<0.002	<0.002
Si	0.24	0.41	0.29	0.48	0.4	0.08	0.15
Sn	0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Ta	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
Ti	<0.01	<0.01	0.08	<0.01	<0.01	0.35	1.36
V	0.18	0.2	0.02	0.05	0.05	0.03	0.01
W	0.15	1.6	1.9	0.01	<0.01	0.13	<0.01
Zr	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.02
As	0.0038	0.0029	0.0019	0.0025	0.0021	0.0002	<0.0001
Bi	<0.00001	<0.00001	<0.00001	0.0008	0.00006	0.00007	0.00017
Pb	0.00005	<0.00001	0.00007	-	-	-	-
Sb	0.00077	0.00041	0.0001	-	-	-	-
C	0.08	0.12	0.004	0.043	0.066	0.091	0.024
S	0.001	0.001	0.002	0.002	0.001	<0.001	0.002
O	0.0032	0.0037	0.0032	0.0032	0.0016	0.0005	0.0006
N	0.0447	0.0359	0.017	0.0604	0.238	0.0065	0.004