

Heat exchangers, Materials and Application to solar and Nuclear power



Mark Anderson

focus on one particular technology gap of sCO2 systems which you think could be resolved or developed further by researchers at university. In other words, the vision of the panel "What Universities can do (or are actually doing) to help make sCO2 a commercial reality?".

heat exchangers/materials and application to nuclear power (i.e., ASME section 3).

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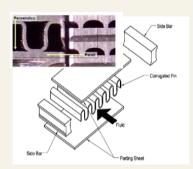


Advanced compact HX for sCO₂

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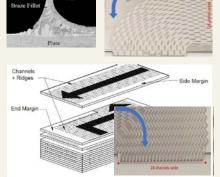
Optimization of HX geometries for f&j

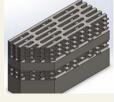
- Need to qualify for ASME Boiler and Pressure Vessel code.
 Section III – nuclear service
- 2. Understand and determine failure mechanisms
- 3. Optimize for low DP and high effectiveness
- 4. Reduction in materials and construction costs.





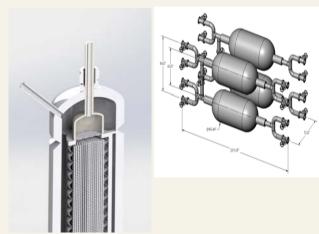




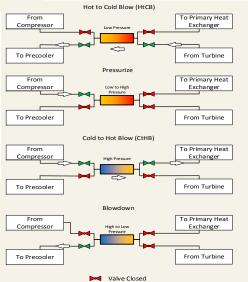




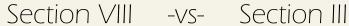




Fixed bed regenerator











VIII - Division 2 (non-Nuclear)

PCHE code case exists

- The most conservative case for non-nuclear applications
- Analysis can be carried out over an entire structure without the need to categorize stresses
 - Limits are imposed uniformly on all points of stress

Plastic collapse

- Stress beyond the yield point is allowed as long as plasticity is appropriately modeled.
- Plasticity models can vary in conservativeness from bilinear to full multilinear implementation of the σ - ϵ curve

Local failure

· Limits are imposed on the extent of plastic strain

Collapse from buckling

 Buckling analysis must be performed on any structures found to be compressively loaded

· Fatigue failure from cyclic loading

- Cyclic loads such as startup/shutdown and load following must be accounted for.
- Implements cycle limits on periodically varying loads.

Organizations Involved

MPR Associates
CompRex, LLC.
Vacuum Process Engineering
Georgia Institute of Technology
North Carolina State University
University of Idaho
University of Michigan
University of Wisconsin
Electric Power Research Institute
Sandia National Laboratories
Phoenix (Nuclear Laboratory), LLC.

III - Division 1 (Nuclear service)

PCHE code case in progress

 Required for any Class 1 components. Metallic vessels, heat exchangers, pumps, piping, valves, etc. used in Nuclear power plants.

Stresses found during analysis have to be classified

- Different limits are applied based on the stress classification
- General primary membrane $P_{\rm m}$, local primary membrane $P_{\rm L}$, primary bending $P_{\rm b}$, expansion $P_{\rm e}$, secondary Q, peak F.

Service level must be specified

- Level A is temperatures and conditions below the onset of creep
- Level B is temperatures where creep occurs; here time limits are imposed based on calculation of creep life
- Level C is temperatures and conditions supporting ratcheting at extreme fatigue. Cycle limits are imposed.

Plasticity

 Strain hardening cannot be counted in models. Only simple elasticperfectly plastic models can be used. This is more conservative than Section VIII.

Local Failure

 Limits on strain are imposed based on stress classification and service level. Service levels B and C allow substantial strain to account for creep and ratcheting.

Buckling

 Buckling analysis must be performed on any structures found to be compressively loaded

Creep

• Creep life of Level B components is evaluated

Fatigue and Ratcheting failure from cyclic loading

- Fatigue and Ratcheting are considered for Level C components
- Fatigue excursions with cycle limits < 10⁶ cycles are not allowed





Code & Commercialization gaps

Section III PCHE Code Case Gaps	Commercialization Gaps
Stress classification rules (Primary, secondary, peak)	Roadmap to Section III certification
Allowable stress limits in diffusion bonded materials	Creep-fatigue quantification methods
Allowable stress and material properties in weldments	Acceptable thermal ramp rate
Determine if heat treatment is required after bonding	Detection methods of fouling and channel plugging
Suitability of existing welding rules for header attachment	Cleaning methods to mitigate scaling and plugging
Examination methods of weld and diffusion—bonded core	Determine limits for cyclical operation
Modify proof pressure testing procedure if necessary	Estimate regular inspection costs
Provide rules for inelastic analysis methods	Special limitations for reactive coolants
Acceptable plastic strains in flow passage region	Utility and requirement of instrumentation
Creep-fatigue curves for diffusion bonded materials	Identify operational quirks using molten metal or salts
Isochronous stress-strain curves	Platform for testing instrumentation
Identify and mitigate all failure modes	FEA Methodology for Section III certification

<u>Developments on PCHE Code</u> <u>Qualification</u>

2005 – requirements for diffusionbonded microchannel heat exchangers outlined in Code Case 2437-1.

2009 – Code Case 2621-1 provided design, fabrication, and inspection requirements. Limited to 304L, 316L, and 2205 stainless.

2011 – Diffusion-bonding (diffusion-welding) was added to allowed Section IX welding processes.

2015 – Nestell and Sham publish "ASME Code Considerations for the Compact Heat Exchanger."

2017 – IRP Grant rewarded for Section III Code Case development

Ongoing – Section III, Division 5 qualification effort of Alloy 617 and 230

Three investigation strategies:

- 1) Finite Element Analysis (EPP, Inelastic)
- 2) Testing of small diffusion-bonded specimen (tensile, creep, creep fatigue)
- 3) Testing of lab-scale PCHEs using a variety of coolants



Printed-Circuit Heat Exchanger (PCHE)



- **O.** Steady State performance obtain Darcy and Colburn factors
 - Are existing flow and heat transfer correlations valid for exotic coolants?
- 1. Creep Test high temperature, high pressure run for 500+ hours on under-designed geometry
 - Where will maximum creep occur? Are creep properties similar to the base material?
- 2. Ratcheting Test subject unit to temperature oscillation for ~1000 cycles
 - When and where will ratcheting occur and will it cause shim separation?
- 3. Thermal Fatigue Test high temperature, moderate pressure
 - Where are cracks most likely to form? How can crack propagation be mitigated?
- 4. Thermal Ramp Test test a Section VIII design under rapid transients
 - How fast can PCHEs be brought up to temperature? What are the load-following limits?
- **5. Fouling/Clogging** measure accumulation in channels and try cleaning methods
 - How can fouling be measured and mitigated? How does this vary with respect to coolant?



Herringbone (zig-zag) Geometry



ShimRex Geometry



Airfoil-fin Geometry

Materials Studied: Alloy 800H and SS316H

The allowable stresses for operation in the creep range are time-dependent and provided in Division 5 for five alloys: 2-1/4Cr-1Mo ferritic steel, Type 304H stainless steel, Type 316H stainless steel, Alloy 800H, and modified 9Cr-Mo-V [Grade 91] ferritic steel. Soon Alloy 617 will be included as well, since it is a code case currently being balloted. Alloy 718 is available in Division 5 for high temperature bolting.



NDE and DE techniques for evaluating CHX

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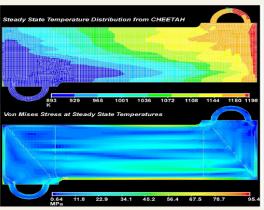






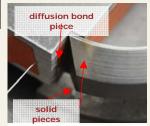
Destructive testing

- -fusion welds
- -diffusion welds
- -dissimilar materials



Thermal performance

Weld Inspection

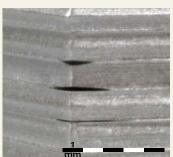




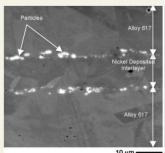
Before welding

Header welds

Diffusion Bond Inspection

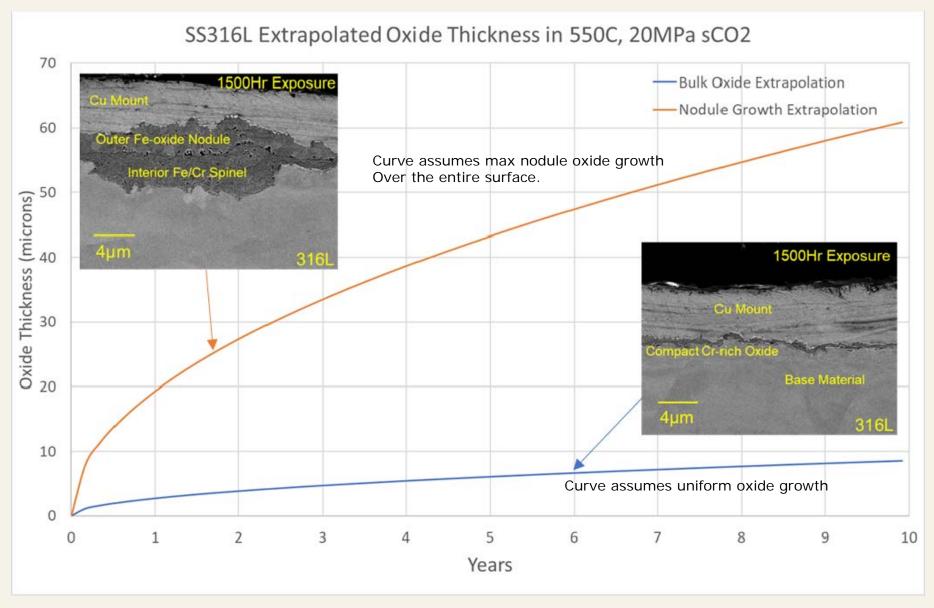




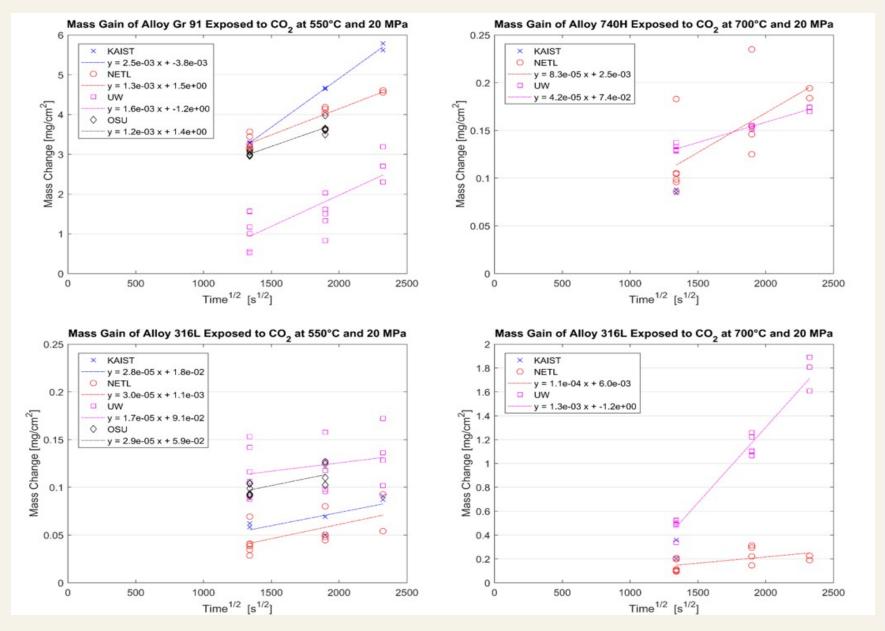


Bond inclusions in alloy 617 [1]

Assessment of pipe/component corrosion allowance



4/11/2018



Paper 146 Tucker, et al 2018, sCO2 symposium (ROUND ROBIN TESTS at multiple institutions)



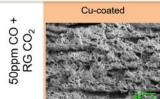
Evaluation of materials, Pipe welds, dissimilar materials welds, mechanical, tensile, corrosion rates, impurities

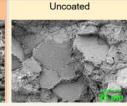












	UTS (MPa)	0.2% YS (MPa)	Elongation (%)
316L: Unexposed	605 ± 1	297 ± 3	81.2 ± 1.2
316L: 750°C CO ₂	598 ± 15	220 ± 1	26.8 ± 4.7
316L: 750°C CO ₂ + 50ppm CO	498 ± 14	291 ± 34	7.5 ± 1.7
316L Cu: 750°C CO ₂	590 ± 18	233 ± 12	29.1 ± 11.0
316L Cu: 750°C CO ₂ + 50ppm CO	641 ± 4	263 ± 4	68.7 ± 4.2
316L-316L: Unexposed	541 ± 5	278 ± 2	41.2 ± 0.8
316L-316L: 750°C CO ₂	586 ± 11	240 ± 13 19.2 ± 0.4	
316L-316L: 750°C CO ₂ + 50ppm CO	538 ± 2	252 ± 1	14.3 ± 0.5
316L-316L Cu: 750°C CO₂	581 ± 21	237 ± 6	21.9 ± 5.4
316L-316L Cu: 750°C CO ₂ + 50ppm CO	651 ± 15	256 ± 4.9	40.1 ± 4.3

Alloy	Oxidation Notes	Oxide Thickness* [microns/1000hr]	Susceptibility to Impurities Corrosion	
T92	Outer Fe rich oxide Inner Fe,Cr rich oxide Spallation observed at 550C	450C - 6.8 550C - 20.3	NA	
T122	Outer Fe rich oxide Inner Fe,Cr rich oxide Spallation observed at 550C	450C - 6.4 550C - 1.8	NA	
347	Outer Fe rich oxide Inner Cr rich oxide Spallation observed at 650/750C	450C2 550C - 1.6 650C2 750C - 5.2	Oxygen rich environments accelerate oxidation/spallation	
316	Cr rich oxide Fe rich nodules with inner Fe,Cr oxide Spallation observed at 650/700C	450C2 550C7 650C - 5.9 700C - 5.8	NA	
310	Cr rich oxide	450C2 550C3 650C7 750C - 1.1	NA	
709	Cr rich oxide	650C7	NA	
800	Cr rich oxide	450C1 550C4 650C8	Oxygen rich environments severely increase oxidation	
718	Cr rich oxide Nodules of Fe rich oxide at T=750C Ti/Al internal oxidation	450C1 550C6 650C8 750C - 1.9	Oxygen rich environments accelerate oxidation/spallation	
625	Cr rich oxide	450C1 550C3 650C3	Oxygen environments enhanced oxidation	
023	GI Hell balde	700C – 1.1 750C – 1.7	CO environments enhanced oxidation and caused oxide buckling	
282	Cr rich oxide Ti/Al internal oxidation Some oxide cracking	Poor resistance to oxy environments 750C – 2.9 Excellent resistance to environments		
617	Cr rich oxide Ti/Al internal oxidation	450C1 550C3	Good resistance to oxygen rich environments	
		650C - 1.2 750C - 1.7	NA	
740	Cr rich oxide Cr/Ti rich nodules oxide nodules	650C5 700C9	Good resistance to oxygen rich environments	
	Ti/Al internal oxidation	750C – 1.7	NA	
230	Cr rich oxide	450C1 550C3 650C3 750C7	Excellent resistance to impurities corrosion	

*Oxide thickness is converted from mass to thickness using density of chromium (spallation and nodule formation can alter these measurements slightly) \sim Estimates are 20 \pm 9% accurate based on SEM imaging of 5 different alloys

This table is based on Research Grade (99.999%) CO₂ with impurities consisting of 100ppm O₂ and 1%CO additions





- Universities have a lot of resources facilities, labor, creative ideas.
- Every project that is done in this area is a collaboration between multiple universities, laboratories, companies.
- Universities are great at disseminating information and building collaborations.

Summary of papers at conference

- 45 Mechanical and Corrosion Performance of the Weld of 740H and 282, Andrew Brittan, University of Wisconsin, Madison
- 59 Switched Bed Regenerators for sCO2 Cycles, Jack Hinze, University of Wisconsin, Madison
- 114 Nuclear Code Case Development of Printed-Circuit Heat Exchangers with Thermal and Mechanical Performance Testing, Shaun Aakre, University of Wisconsin, Madison
- 130 Thermal-Hydraulic Performance of Compact Diffusion Bonded Heat Exchanger Geometries Using Supercritical Carbon Dioxide as the Working Fluid, Sandeep R. Pidaparti, Georgia Institute of Technology
- 144 Supercritical Brayton Power Conversion with a Direct Cooled Reactor for Space Power, Becky Sondelski, University of Wisconsin, Madison
- 148 Processing and Properties of Robust Ceramic/Metal Composites for Heat Exchangers Operating at >750°C with Supercritical CO2, Kenneth Sandhage, Purdue University
- 146 Supercritical CO2 Round Robin Test Program, Julie Tucker, Oregon State University